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- (71) Applicant (*for all designated States except US*): **OUT-OKUMPU OYJ** [FI/FI]; Riihitontuntie 7, FIN-02200 Espoo (FI).
- (72) Inventors; and
- (75) Inventors/Applicants (*for US only*): **HIRSCH, Martin** [DE/DE]; Am Vogelschutz 5, 61381 Friedrichsdorf (DE). **STRÖDER, Michael** [DE/DE]; Dürerstrasse 77, 61267 Neu-Anspach (DE). **WEBER, Peter** [DE/FI]; Kuninkaanlahdentie 6C, FIN-02160 Espoo (FI).
- (74) Agent: **KEIL & SCHAAFHAUSEN**; Cronstettenstrasse 66, 60322 Frankfurt am Main (DE).
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(54) Title: METHOD AND PLANT FOR THE HEAT TREATMENT OF SOLIDS CONTAINING IRON OXIDE

(57) Abstract: A method for the heat treatment of solids containing iron oxide, in which fine-grained solids are heated to a temperature of about 630°C in a fluidized-bed reactor (1). To improve the utilization of energy, it is proposed to introduce a first gas or gas mixture from below through a supply tube (3) into a mixing chamber (7) of the reactor (1), the gas supply tube (3) being at least partly surrounded by a stationary annular fluidized bed (10) which is fluidized by supplying fluidizing gas. The gas velocities of the first gas or gas mixture and of the fluidizing gas for the annular fluidized bed (10) are adjusted such that the Particle-Froude-Numbers in the gas supply tube (3) are between 1 and 100, in the annular fluidized bed (10) between 0.02 and 2, and in the mixing chamber (7) between 0.3 and 30.

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5 **METHOD AND PLANT FOR THE HEAT TREATMENT**
 OF SOLIDS CONTAINING IRON OXIDE

Technical Field

10 The present invention relates to a method for the in particular reductive heat treatment of solids containing iron oxide, in which fine-grained solids are heated to a temperature of about 630°C in a fluidized-bed reactor, and to a corresponding plant.

15 Such method and a plant are known for instance from DE 44 10 093 C1, in order to reduce iron-oxide-containing solids such as iron ores, iron ore concentrates or the like. For this purpose, iron-oxide-containing ore is introduced into the fluidized-bed reactor and fluidized with heated reduction gas. The solids are entrained by the gas stream and separated from the exhaust gas in a downstream separator, in order to be recirculated to the reactor. For further processing, solids are withdrawn from the lower region of the reactor.

20 However, this direct reduction in the fluidized-bed reactor involves the risk that above the distributor grid (gas distributor), which is typically used for distributing the reduction gas, excess temperatures are obtained, as the reduction gas has a high temperature for heating the solids. At the same time, the mass and heat transfer conditions are re-
25 garded as unsatisfactory due to the only limited solids retention time of e.g. 20 minutes and the only moderate relative velocities between gas and solids. Due to the additional introduction of exhaust gases of a downstream reduction stage into the reactor, the reactor must have a complicated design, for instance with a diameter reduced by about 30 % in the vicinity of the exhaust gas inlet. In addition, the gas supply conduits are
30 gated at an angle, in order to prevent a clogging of the conduit to which dust-laden (secondary) gas is admitted, and to allow that solids which have entered the conduit fall back into the reactor.

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Description of the Invention

Therefore, it is the object of the present invention to improve the conditions for a mass and heat transfer during the heat treatment of solids containing iron oxide.

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In accordance with the invention, this object is solved by a method as mentioned above, in which a first gas or gas mixture is introduced from below through at least one preferably centrally arranged gas supply tube (central tube) into a mixing chamber region of the reactor, the central tube being at least partly surrounded by a stationary annular fluidized bed which is fluidized by supplying fluidizing gas, and in which the gas velocities of the first gas or gas mixture as well as of the fluidizing gas for the annular fluidized bed are adjusted such that the Particle-Froude-Numbers in the central tube are between 1 and 100, in the annular fluidized bed between 0.02 and 2 and in the mixing chamber between 0.3 and 30.

15

In the method of the invention, the advantages of a stationary fluidized bed, such as a sufficiently long solids retention time, and the advantages of a circulating fluidized bed, such as a good mass and heat transfer, can surprisingly be combined with each other during the heat treatment, while the disadvantages of both systems are avoided. When passing through the upper region of the central tube, the first gas or gas mixture entrains solids from the annular stationary fluidized bed, which is referred to as annular fluidized bed, into the mixing chamber, so that due to the high slip velocities between the solids and the first gas an intensively mixed suspension is formed and an optimum mass and heat transfer between the two phases is achieved. By correspondingly adjusting the bed height in the annular fluidized bed as well as the gas velocities of the first gas or gas mixture and of the fluidizing gas, the solids loading of the suspension above the orifice region of the central tube can be varied within wide ranges, so that the pressure loss of the first gas between the orifice region of the central tube and the upper outlet of the mixing chamber can be between 1 mbar and 100 mbar. In the case of a high solids loading of the suspension in the mixing chamber, a large part of the solids will separate out of the suspension and fall back into the annular fluidized bed. This recirculation is called internal solids recirculation, the stream of solids circulating in this internal circulation normally being significantly larger than the amount of solids supplied

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to the reactor from outside. The (smaller) amount of not precipitated solids is discharged from the mixing chamber together with the first gas or gas mixture. The retention time of the solids in the reactor can be varied within a wide range by the selection of height and cross-sectional area of the annular fluidized bed and be adapted to the desired heat treatment. Due to the high solids loading on the one hand and the good mass and heat transfer on the other hand, the formation of local temperature peaks in the mixing chamber can be avoided. The amount of solids entrained from the reactor with the gas stream is completely or at least partly recirculated to the reactor, with the recirculation expediently being fed into the stationary fluidized bed. The stream of solid matter thus recirculated to the annular fluidized bed normally lies in the same order of magnitude as the stream of solid matter supplied to the reactor from outside. Apart from the excellent utilization of the reduction gas, another advantage of the method in accordance with the invention consists in the possibility of quickly, easily and reliably adjusting the utilization of the reduction gas and the mass transfer to the requirements by changing the flow velocities of the first gas or gas mixture and of the fluidizing gas. Furthermore, the construction of the reactor can be simplified such that the same for instance has a cylindrical shape.

To ensure a particularly effective heat transfer in the mixing chamber and a sufficient retention time in the reactor, the gas velocities of the first gas mixture and of the fluidizing gas are preferably adjusted for the fluidized bed such that the dimensionless Particle-Froude-Numbers (Fr_p) in the central tube are 1.15 to 20, in particular about 10.6, in the annular fluidized bed 0.115 to 1.15, in particular about 0.28, and/or in the mixing chamber 0.37 to 3.7, in particular about 1.1. The Particle-Froude-Numbers are each defined by the following equation:

$$Fr_p = \frac{u}{\sqrt{\frac{(\rho_s - \rho_f)}{\rho_f} * d_p * g}}$$

with

- 30 u = effective velocity of the gas flow in m/s
 ρ_s = density of a solid particle in kg/m³

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ρ_f = effective density of the fluidizing gas in kg/m³
 d_p = mean diameter in m of the particles of the reactor inventory (or the particles formed) during operation of the reactor
 g = gravitational constant in m/s².

5

When using this equation it should be considered that d_p does not indicate the mean diameter (d_{50}) of the material used, but the mean diameter of the reactor inventory formed during the operation of the reactor, which can differ significantly from the mean diameter of the material used (primary particles). Even from very fine-grained material with a mean diameter of e.g. 3 to 10 μm , particles (secondary particles) with a mean diameter of 20 to 30 μm can be formed for instance during the heat treatment. On the other hand, some materials, e.g. ores, are decrepitated during the heat treatment.

10

In accordance with a development of the invention it is proposed to adjust the bed height of solids in the reactor such that the annular fluidized bed at least partly extends beyond the upper orifice end of the central tube by a few centimeters, and thus solids are constantly introduced into the first gas or gas mixture and entrained by the gas stream to the mixing chamber located above the orifice region of the central tube. In this way, there is achieved a particularly high solids loading of the suspension above the orifice region of the central tube, which improves the transfer conditions between gas and solids.

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By means of the method in accordance with the invention all kinds of iron-oxide-containing ores, in particular iron ores or iron ore concentrates, can effectively be heat-treated.

25

The generation of the amount of heat necessary for the operation of the reactor can be effected in any way known to the expert for this purpose.

30

In accordance with a particular embodiment of the invention it is provided to supply preheated reduction gas to the reactor for fluidization, which reduces the possibly likewise preheated solids. The reactor temperature for instance lies below the temperature

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of the mass flows entering the reactor. As reduction gas, gas with a hydrogen content of at least 80 %, preferably above 90 %, is particularly useful.

5 The consumption of fresh reduction gas can be decreased considerably when the reduction gas is cleaned in a reprocessing stage downstream of the reactor and subsequently recirculated to the reactor. During reprocessing, the gas is first separated from solids, possibly passed through a scrubber and cooled below the dew point of steam, so that the steam content can be reduced, then compressed and enriched with fresh hydrogen.

10 In accordance with a development of the invention it is proposed to cover part of the energy demand of the reactor by supplying exhaust gases from a downstream reactor, for instance another reduction reactor, which possibly still contains reduction gas. Thus, the necessary demand of fresh reduction gas and fuel can be decreased distinctly.
15 Preferably, the exhaust gas is supplied to the reactor via the central tube, while processed reduction gas is expediently introduced as fluidizing gas into the annular fluidized bed through a conduit.

20 A plant in accordance with the invention, which is in particular suited for performing the method described above, has a reactor constituting a fluidized-bed reactor for the in particular reductive heat treatment of solids containing iron oxide, the reactor having a gas supply system which is formed such that gas flowing through the gas supply system entrains solids from a stationary annular fluidized bed, which at least partly surrounds the gas supply system, into the mixing chamber. Preferably, this gas supply
25 system extends into the mixing chamber. It is, however, also possible to let the gas supply system end below the surface of the annular fluidized bed. The gas is then introduced into the annular fluidized bed e.g. via lateral apertures, entraining solids from the annular fluidized bed into the mixing chamber due to its flow velocity.

30 In accordance with a preferred aspect of the invention, the gas supply system has a central tube extending upwards substantially vertically from the lower region of the reactor into the mixing chamber of the reactor, which central tube is at least partly surrounded by a chamber in which the stationary annular fluidized bed is formed. The an-

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nular fluidized bed need not be circular, but rather other shapes of the annular fluidized bed are also possible in dependence on the geometry of the central tube and of the reactor, as long as the central tube is at least partly surrounded by the annular fluidized bed. Of course, two or more central tubes with different or the same dimensions can
5 also be provided in the reactor. Preferably, however, at least one of the central tubes is arranged approximately centrally, based on the cross-sectional area of the reactor.

In accordance with another embodiment of the present invention, the central tube has apertures, for instance in the form of slots, at its shell surface, so that during the opera-
10 tion of the reactor solids constantly get into the central tube through the apertures and are entrained by the first gas or gas mixture from the central tube into the mixing chamber.

In accordance with a preferred embodiment, a separator, for instance a cyclone for
15 separating solids is provided downstream of the reactor, the separator having a solids conduit which leads to the annular fluidized bed of the first reactor.

To provide for a reliable fluidization of the solids and the formation of a stationary fluid-
20 ized bed, a gas distributor is provided in the annular chamber of the reactor, which divides the chamber into an upper fluidized bed region and a lower gas distributor chamber. The gas distributor chamber is connected with a supply conduit for fluidizing gas. Instead of the gas distributor chamber, there can also be used a gas distributor composed of tubes.

The energy demand of the plant can be reduced in that the reactor has a supply con-
25 duit for hydrogen-containing reduction gas, which leads to the central tube and is connected for instance with the exhaust gas outlet of a separator of another reactor provided downstream of the reactor. Alternatively or in addition, a supply conduit for pre-heated hydrogen-containing reduction gas, which extends in or leads to the annular
30 chamber, may be provided in the plant in accordance with the invention.

To not exceed the technically manageable maximum temperatures of the gases used in the plant, the energy required for the heat treatment should preferably not exclusively

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be introduced into the reactor via the gases. For this purpose, a preheating stage for the solids may be provided upstream of the reactor, so that already preheated solids are introduced into the reactor. Preferably, the temperature of the solids charged into the reactor lies above the reactor temperature.

5

In the annular fluidized bed and/or the mixing chamber of the reactor, means for deflecting the solid and/or fluid flows may be provided in accordance with the invention. It is for instance possible to position an annular weir, whose diameter lies between that of the central tube and that of the reactor wall, in the annular fluidized bed such that the upper edge of the weir protrudes beyond the solids level obtained during operation, whereas the lower edge of the weir is arranged at a distance from the gas distributor or the like. Thus, solids separated out of the mixing chamber in the vicinity of the reactor wall must first pass by the weir at the lower edge thereof, before they can be entrained by the gas flow of the central tube back into the mixing chamber. In this way, an exchange of solids is enforced in the annular fluidized bed, so that a more uniform retention time of the solids in the annular fluidized bed is obtained.

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The invention will subsequently be described in detail with reference to an embodiment and the drawing. All features described and/or illustrated in the drawing form the subject-matter of the invention per se or in any combination, independent of their inclusion in the claims or their back-reference.

Brief Description of the Drawings

25 Fig. 1 shows a process diagram of a method and a plant in accordance with an embodiment of the present invention; and

Fig. 2 shows an enlarged detail of Fig. 1.

30

Detailed Description of the Preferred Embodiments

In the method shown in Figure 1, which is in particular suited for the heat treatment of solids containing iron oxide, solids are introduced into a reactor 1 via a supply conduit

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2, as can be seen in the enlarged representation of Figure 2. The for instance cylindrical reactor 1 has a central tube 3 arranged approximately coaxially with the longitudinal axis of the reactor, which central tube extends substantially vertically upwards from the bottom of the reactor 1.

5

In the vicinity of the bottom of the reactor 1, an annular gas distributor chamber 4 is provided, which at its upper end is terminated by a gas distributor 5 having through openings. A supply conduit 6 opens into the gas distributor chamber 4.

10

In the vertically upper region of the reactor 1, which forms a mixing chamber 7, a discharge conduit 8 is arranged, which opens into a separator 9 constituting a cyclone.

15

When solids are now introduced into the reactor 1 via the supply conduit 2, a layer annularly surrounding the central tube 3 is formed on the gas distributor 5, which layer is referred to as annular fluidized bed 10. Fluidizing gas introduced into the gas distributor chamber 4 through the supply conduit 6 flows through the gas distributor 5 and fluidizes the annular fluidized bed 10, so that a stationary fluidized bed is formed. The velocity of the gases supplied to the reactor 1 via the gas distributor chamber 4 is adjusted such that the Particle-Froude-Number in the annular fluidized bed 10 is about 0.28.

20

25

Due to the supply of more solids into the annular fluidized bed 10, the solids level 11 in the reactor 1 rises to such an extent that solids get into the orifice of the central tube 3. Through the central tube 3, a gas or gas mixture is at the same time introduced into the reactor 1. The velocity of the gas supplied to the reactor 1 preferably is adjusted such that the Particle-Froude-Number in the central tube 3 is about 10.6 and in the mixing chamber 7 about 1.1. Due to these high gas velocities, the gas flowing through the central tube 3 entrains solids from the stationary annular fluidized bed 10 into the mixing chamber 7 when passing through the upper orifice region.

30

Since the level 11 of the annular fluidized bed 10 is raised above the upper edge of the central tube 3, solids flow over this edge into the central tube 3, whereby an intensively mixed suspension is formed. The upper edge of the central tube 3 can be straight, wavy or serrated, or the shell surface can have lateral inlet openings. As a result of the

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reduction of the flow velocity due to the expansion of the gas jet and/or by impingement on one of the reactor walls, the entrained solids quickly lose velocity and partly fall back into the annular fluidized bed 10. The amount of non-precipitated solids is discharged from the reactor 1 together with the gas stream via conduit 8. Between the reactor regions of the stationary annular fluidized bed 10 and the mixing chamber 7 a solids circulation is obtained, by means of which a good heat transfer is ensured. Before the further processing, the solids discharged via conduit 8 are separated from the gases or gas mixtures in the cyclone 9.

In accordance with the method shown in Fig. 1, the fine-grained solids such as iron ore are first charged into a preheating stage with a Venturi preheater 12. Downstream of the same a cyclone 13 is provided, in which the solids are separated from exhaust gas. From the cyclone 13, the solids are supplied to another Venturi preheater 14. Downstream of the same, a cyclone 15 is in turn provided, in which the solids are separated from exhaust gas and via a bunker 16 and a screw conveyor 17 are supplied to the reactor 1 via conduit 2.

Hot combustion gases from a combustion chamber 18 are supplied to the Venturi preheater 14 for heating the solids, to which combustion chamber fuel is supplied via conduit 10 and combustion air is supplied via conduit 20. It turned out to be advantageous to operate the combustion at a pressure of 0.8 to 10 bar and preferably at atmospheric pressure. The still hot exhaust gases which were separated from the solids in the cyclone 15 are supplied to the first Venturi preheater 12 for preheating the solids. After the separation of the solids in the cyclone 13, the exhaust gas is cleaned in a filter 21.

In the reactor 1, the solids are subjected to a heat treatment, with heated reducing fluidizing gas being introduced through conduit 6 into the annular fluidized bed 10 formed. At the same time, exhaust gas from another reduction stage downstream of the reactor 1 is supplied through the central tube 3, so that the solids for one part circulate in the reactor 1 in the manner described above and for the other part are discharged from the reactor 1 via conduit 8 and upon separation of the exhaust gas in the cyclone 9 are recirculated to the annular fluidized bed via conduit 22.

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From the annular fluidized bed 10, a stream of solids is in addition withdrawn from the reactor 1 and supplied to a downstream fluidized bed reactor 23. The fluidized-bed reactor 23 has a classical fluidized bed, into which heated fluidizing gas is introduced via conduit 24. Via conduit 25, solids are withdrawn from the fluidized-bed reactor 23 and
5 supplied for instance to a briquetting plant 26.

In a cyclone 27, the exhaust gases of the fluidized-bed reactor 23 are separated from solids, which are recirculated to the fluidized-bed reactor 23 via conduit 28. The exhaust gases are supplied from the cyclone 27 via conduit 29 to the central tube 3 of the
10 reactor 1.

The exhaust gases of the reactor 1, which were separated from the solids in the cyclone 9, are supplied to a reprocessing via conduit 30. First of all, the exhaust gases are cooled in a heat exchanger 31 and introduced into a scrubber 32, where the cooled
15 gas is further cooled below the dew point of steam, so that the steam content of the exhaust gas can largely be removed. Via conduit 33, a partial stream of the exhaust gas can be removed from the circuit, to prevent for instance an accumulation of nitrogen in the circulating gas. In the same way, fresh reducing gas can be admixed via
20 conduit 34 for fortification. The cleaned gas is now preheated in the heat exchanger 31 and supplied to a heater 35. The cleaned hot reduction gas is then supplied to the fluidized-bed reactor 23 via conduit 24 and as fluidizing gas via conduit 6 to the reactor 1.

In the following, the invention will be explained by means of an example demonstrating the invention, but not restricting the same.

25

Example (Reduction of iron-oxide-containing iron ore)

In a plant corresponding to Fig. 1, 61.2 t/h moist ore with 7.8 % moisture were supplied to the Venturi preheater 12. Into the combustion chamber 18, 1500 Nm³/h natural gas
30 were at the same time introduced via conduit 19 and 21.000 Nm³/h air via conduit 20. By means of the combustion gases from the combustion chamber 18, the moist ores were preheated to 500°C in the two Venturi preheaters 12 and 14. In the filter 21, 2.6 t/h dust were separated from the exhaust gas of the preheating stage.

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To the reactor 1, 54.2 t/h preheated ore were supplied via the screw conveyor 17 and conduit 2, and reducing fluidizing gas containing

5 91 vol-% H₂,
 0.6 vol-% H₂O and
 8.4 vol-% N₂

was supplied via conduit 6. The reduction temperature in the reactor 1, which had a
10 diameter of 3 m, was about 630°C. The pressure at the outlet of the reactor 1 was 4
bar.

From the reactor 1, 40.6 t/h prereduced material with a degree of metallization of 70 %
were introduced into the fluidized-bed reactor 23. The same had a length of 12 m and a
15 width of 4 m. To the fluidized-bed reactor 23, preheated reducing fluidizing gas contain-
ing

 91 vol-% H₂,
 0.6 vol-% H₂O and
20 8.4 vol-% N₂

was supplied via conduit 24.

From the fluidized-bed reactor 23, 36.8 t/h product with a degree of metallization of 92
25 % were introduced via conduit 25 into the briquetting plant 26 and briquetted therein.
The product had a carbon content of 0.05 wt-%.

Via conduit 30, 182,000 Nm³/h exhaust gas containing

30 79 vol-% H₂,
 12 vol-% H₂O, and
 9 vol-% N₂

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were introduced into the heat exchanger 31 and cooled therein to 120°C. In the scrubber 32, the exhaust gas was further cooled to 28°C. Upon admixture of 23,000 Nm³/h fresh gas with a H₂ content of 97 % via conduit 34, the gas was heated to 520°C in the heat exchanger 31. After a further heating in the heater 35, 70 % of the gas were introduced into the fluidized-bed reactor 23, and the remaining 30 % of the gas were introduced into the reactor 1.

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List of reference numerals:

	1	reactor	19	conduit
	2	supply conduit	20	conduit
5	3	central tube	21	filter
	4	gas distributor chamber	22	solids return conduit
	5	gas distributor	23	fluidized-bed reactor
	6	supply conduit	24	conduit
	7	mixing chamber	25	conduit
10	8	conduit	26	briquetting plant
	9	separator	27	cyclone
	10	annular fluidized bed	28	solids return conduit
	11	level of the annular fluidized bed	29	conduit
	12	Venturi preheater	30	conduit
15	13	cyclone	31	heat exchanger
	14	Venturi preheater	32	scrubber
	15	cyclone	33	conduit
	16	bunker	34	conduit
	17	screw conveyor	35	heater
20	18	combustion chamber		

Claims

1. A method for the heat treatment of solids containing iron oxide, in which fine-grained solids are heated to a temperature of about 450 to 950°C in a fluidized-bed reactor (1), **characterized in that** a first gas or gas mixture is introduced from below into a mixing chamber region (7) of the reactor (1) through a preferably central gas supply tube (3), the gas supply tube (3) being at least partly surrounded by a stationary annular fluidized bed (10) which is fluidized by supplying fluidizing gas, and that the gas velocities of the first gas or gas mixture and of the fluidizing gas for the annular fluidized bed (10) are adjusted such that the Particle-Froude-Numbers in the gas supply tube (3) are between 1 and 100, in the annular fluidized bed (10) between 0.02 and 2, and in the mixing chamber (7) between 0.3 and 30.
2. The method as claimed in claim 1, **characterized in that** the Particle-Froude-Number in the gas supply tube (3) is between 1.15 and 20, in particular about 10.6.
3. The method as claimed in claim 1 or 2, **characterized in that** the Particle-Froude-Number in the annular fluidized bed (10) is between 0.115 and 1.15, in particular about 0.28.
4. The method as claimed in any of the preceding claims, **characterized in that** the Particle-Froude-Number in the mixing chamber (7) is between 0.37 and 3.7, in particular about 1.1.
5. The method as claimed in any of the preceding claims, **characterized in that** the bed height of solids in the reactor (1) is adjusted such that the annular fluidized bed (10) at least partly extends beyond the upper orifice end of the gas supply tube (3) and that solids are constantly introduced into the first gas or gas mixture and are entrained by the gas stream to the mixing chamber (7) located above the orifice region of the gas supply tube (3).

6. The method as claimed in any of the preceding claims, **characterized in that** iron-oxide-containing ore, in particular iron ore or iron ore concentrate is used as starting material.

5 7. The method as claimed in any of the preceding claims, **characterized in that** the fluidizing gas introduced into the annular fluidized bed (10) of the reactor (1) is a preheated reduction gas which contains at least 80 %, in particular more than 90 % hydrogen.

10 8. The method as claimed in claim 7, **characterized in that** the reduction gas is cleaned in a reprocessing stage (31, 32, 33, 34, 35) downstream of the reactor (1) and is subsequently recirculated to the reactor (1).

15 9. The method as claimed in any of the preceding claims, **characterized in that** downstream of the reactor (1) another fluidized-bed reactor (23) is provided, whose exhaust gases are separated from solids in a separator (27) and are introduced into the gas supply tube (3) of the reactor (1).

(20 10. The method as claimed in any of the preceding claims, **characterized in that** upstream of the reactor (1) at least one preheating stage (12, 13, 14, 15) is provided for heating the solids.

25 11. A plant for the heat treatment of solids containing iron oxide, in particular for performing a method as claimed in any of claims 1 to 10, comprising a reactor (1) constituting a fluidized bed reactor, **characterized in that** the reactor (1) has a gas supply system which is formed such that gas flowing through the gas supply system entrains solids from a stationary annular fluidized bed (10), which at least partly surrounds the gas supply system, into the mixing chamber (7).

30 12. The plant as claimed in claim 11, **characterized in that** the gas supply system has at least one gas supply tube (3) which extends upwards substantially vertically from the lower region of the reactor (1) into a mixing chamber (7) of the reactor (1), the gas

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supply tube (3) being at least partly surrounded by an annular chamber in which the stationary annular fluidized bed (10) is formed.

5 13. The plant as claimed in claim 12, **characterized in that** the gas supply tube (3) is arranged approximately centrally with reference to the cross-sectional area of the reactor (1).

10 14. The plant as claimed in any of claims 11 to 13, **characterized in that** the gas supply tube (3) has openings, for instance in the form of slots, at its shell surface.

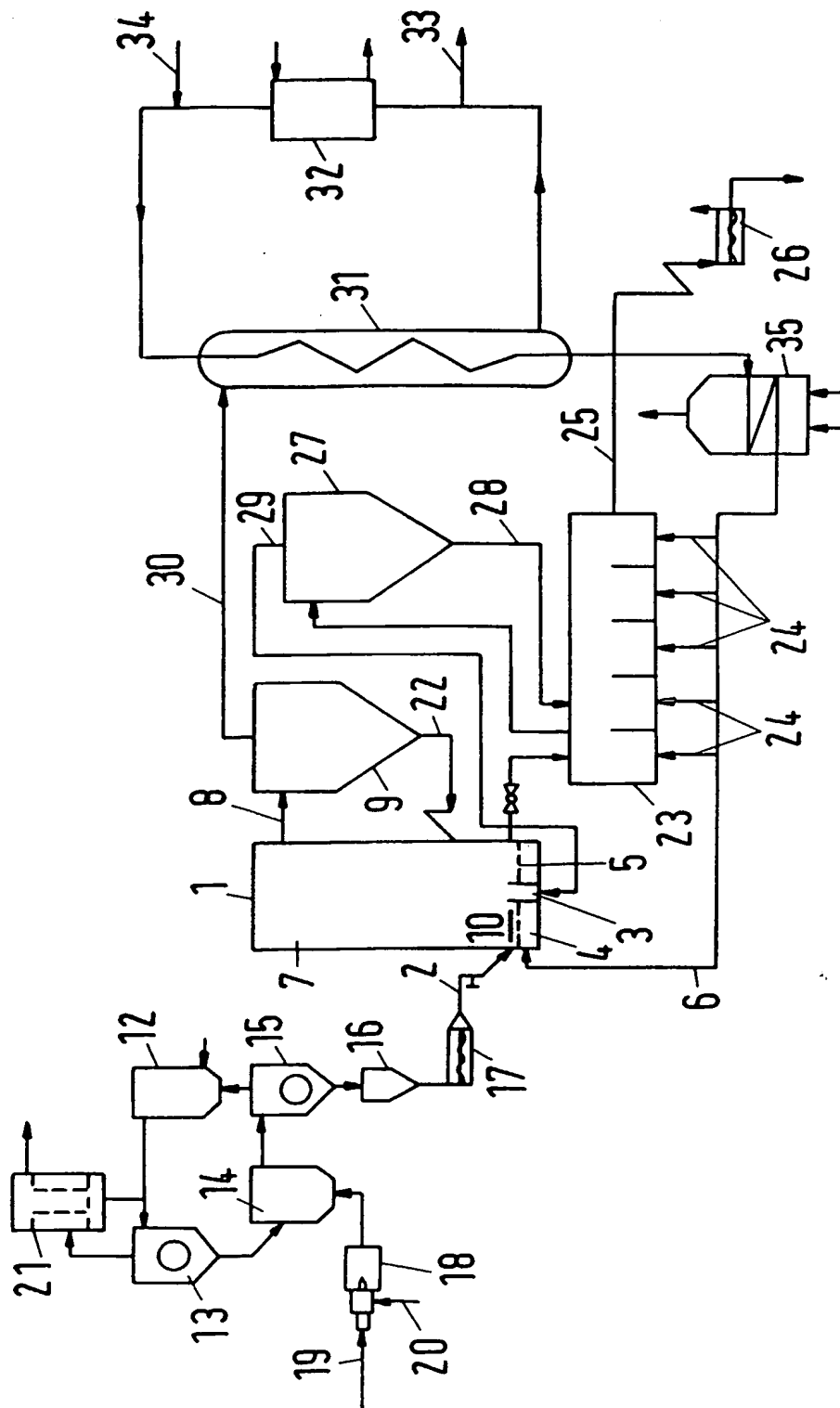
15 15. The plant as claimed in any of claims 11 to 14, **characterized in that** a cyclone (9) for separating solids is provided downstream of the reactor (1), and that the cyclone (9) has a solids conduit (22) leading to the annular fluidized bed (10) of the reactor (1).

20 16. The plant as claimed in any of claims 11 to 15, **characterized in that** in the annular chamber of the reactor (1) a gas distributor (5) is provided, which divides the chamber into an upper fluidized bed region (10) and a lower gas distributor chamber (4), and that the gas distributor chamber (4) is connected with a supply conduit (6) for fluidizing gas.

25 17. The plant as claimed in any of claims 11 to 16, **characterized in that** the reactor (1) has a supply conduit for hydrogen-containing reduction gas, which leads to the gas supply tube (3) and is connected for instance with the exhaust gas outlet of a separator (27) of another reactor (23) downstream of the reactor (1), and/or has a supply conduit for preheated hydrogen-containing reduction gas, which leads to the annular chamber.

30 18. The plant as claimed in any of claims 11 to 17, **characterized in that** a preheating stage (12, 13, 14, 15) for the solids is provided upstream of the reactor (1).

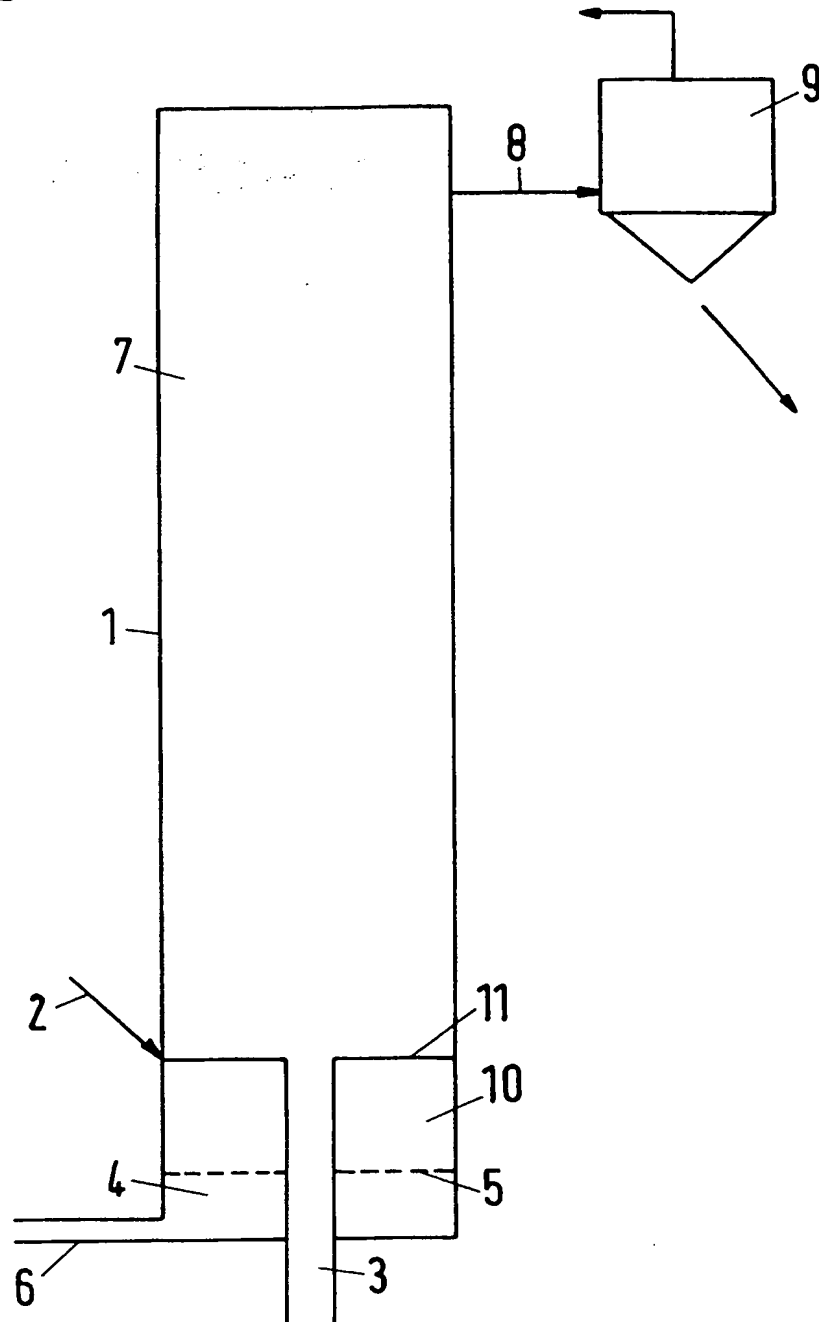
Fig.1



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Fig.2



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INTERNATIONAL SEARCH REPORT

(PCT Article 18 and Rules 43 and 44)

Applicant's or agent's file reference 0 1 P 73 WO	FOR FURTHER ACTION see Notification of Transmittal of International Search Report (Form PCT/ISA/220) as well as, where applicable, item 5 below.	
International application No. PCT/EP 03/14106	International filing date (day/month/year) 12/12/2003	(Earliest) Priority Date (day/month/year) 23/12/2002
Applicant OUTOKUMPU OYJ		

This International Search Report has been prepared by this International Searching Authority and is transmitted to the applicant according to Article 18. A copy is being transmitted to the International Bureau.

This International Search Report consists of a total of 3 sheets.

☒ It is also accompanied by a copy of each prior art document cited in this report.

1. Basis of the report

a. With regard to the language, the international search was carried out on the basis of the international application in the language in which it was filed, unless otherwise indicated under this item.

☐ the international search was carried out on the basis of a translation of the international application furnished to this Authority (Rule 23.1(b)).

b. With regard to any nucleotide and/or amino acid sequence disclosed in the international application, the international search was carried out on the basis of the sequence listing :

☐ contained in the international application in written form.

☐ filed together with the international application in computer readable form.

☐ furnished subsequently to this Authority in written form.

☐ furnished subsequently to this Authority in computer readable form.

☐ the statement that the subsequently furnished written sequence listing does not go beyond the disclosure in the international application as filed has been furnished.

☐ the statement that the information recorded in computer readable form is identical to the written sequence listing has been furnished

2. ☐ Certain claims were found unsearchable (See Box I).

3. ☐ Unity of invention is lacking (see Box II).

4. With regard to the title,

☒ the text is approved as submitted by the applicant.

☐ the text has been established by this Authority to read as follows:

5. With regard to the abstract,

☒ the text is approved as submitted by the applicant.

☐ the text has been established, according to Rule 38.2(b), by this Authority as it appears in Box III. The applicant may, within one month from the date of mailing of this international search report, submit comments to this Authority.

6. The figure of the drawings to be published with the abstract is Figure No.

☒ as suggested by the applicant.

☐ because the applicant failed to suggest a figure.

☐ because this figure better characterizes the invention.

1 _____
☐ None of the figures.

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INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP 03/14106

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C22B5/14 C22B1/10

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C22B B01J C21B

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	DE 44 10 093 C (METALLGESELLSCHAFT AG) 9 March 1995 (1995-03-09) abstract	1-18
A	US 4 806 158 A (HIRSCH MARTIN ET AL) 21 February 1989 (1989-02-21) abstract	1-18
A	US 4 789 580 A (HIRSCH MARTIN ET AL) 6 December 1988 (1988-12-06) abstract	1-18



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search

19 July 2004

Date of mailing of the international search report

09. 08. 2004

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

ANNA-MAJ MAGNUSSON/ELY

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INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP 03/14106

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
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